PREPARATIONS FROM SACCHAROLA

REVISIAE*

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In earlier work from this laboratory, it was shown that fatty acid synthesis in yeast preparations is strongly influenced by carbon dioxide (Klein, 1957) and by a particular fraction sedimentable from postmitochondrial supernatants (Klein, and Booher, 1956; Klein, 1960; Abraham et al., 1961; Den and Klein, 1961; Klein, 1963). Working with animal extracts, other workers have emphasized that citrate and other biochemical intermediates can exert profound effects on fatty acid formation, and several hypotheses based upon such interactions have been formulated (Vagelos et al., 1963; Tubbs and Garland, 1963; Bortz and Lynen, 1963; Vagelos, 1964; Howard and Lowenstein, 1964).

It is the purpose of this report to indicate that yeast preparations are also subject to controlling influences by certain intermediates of the oxidative and fermentative metabolism of glucose.

Saccharomyces cerevisiae, strain IK2G12, was grown and aerated as previously described (Klein, 1957) and disrupted in in interior French pressure cell (Klein, 1963). The high-speed supernatant (H.S.S.) obtained after centrifugation at 85,000 × G for 60 min. was at all overnight against 0.1 M potassium phosphate buffer, pH 6.5, containing 0.5 mM reduced glutathione. This preparation was used in all

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measured.* For determination of the system of the H.S.S. was a second of th

Citrate, which is known to still alar. Navey acid synthesis in animal preparations (Vagelos, 196%), protestably increased the rate of incorporation of acetate into fatty acids (Table I). That citrate does not stimulate by chelation of metals is probable since wersene, over a concentration range of 0.02 mM to 2 mM (where it became inhibitory), did not increase the rate of fatty acid formation. Isocitrate affected the yeast preparation in a manner analogous to citrate, while a-keto-glutarate had no effect. As is the case for the compounds referred to below, citrate and isocitrate did not stimulate the synthesis of nonsaponifiable lipids.

L- α -glycerophosphate was also very effective in stimulating fatty acid synthesis (Table II). Concentrations of the order of 1 mM are evidently optimal in this stimulation. Other related compounds that were found to stimulate in this manner are glucose-6-phosphate and fructose-1.6-diphosphate, both of which might be converted to α -glycerophosphate in these preparations.

Stimulation by L-a-glycerophosphate has been observed by Howard and Lowenstein (1964) in a rat liver system containing supernatant and microsomes. On the basis of their observations, these authors suggested that this compound may act by removing inhibitory amounts of acyl-CoA via glyceride synthesis, since long-chain acyl-CoA compounds were shown

^{*}Methods for isolating major lipid fractions are similar to those used previously (Klein, 1957, 1960), except that following hydrolysis of the incubated suspensions, the hydrolysates were first acidified and extracted with petroleum ether to remove both fatty acids and nonsaponifiable lipids. Analysis by gas chromatography indicated that all the radioactive fatty acids extracted in this manner were 12-18 carbons long, chain lengths of 16 and 18 accounting for 75-80% of the products.

to inhibit fatty acid synthesis in himman and a feature hard and, 1958; Tubbs and Garland, 1963; Bortz and Lynen, A. J. . This hypothesis was tested in our system by incubating palmittal-OcA with H.S.S. plus cofactors in the presence and absence of L- α -glyceropnosphate. Table III illustrates that, although palmityl-CoA severely inhibited fatty acid formation, L-a-glycerophosphate failed to reverse this inhibition. It would appear, therefore, that the mechanism of stimulation by L-a-glycerophosphate in these preparations is not by the reversal of acyl-CoA inhibition. This conclusion is supported by the recent report of Kuhn and Lynen (1965) that, in baker's yeast, the enzyme required for phosphatidic acid synthesis is located exclusively in a particulate fraction of the cell. Table III also shows that palmityl-CoA inhibits the synthesis of nonsaponifiable lipids in addition to fatty acids. Furthermore, the incorporation of malonyl CoA into fatty acids was also found to be impaired by low concentrations of palmityl-CoA, although the incorporation of acetate into fatty acids is more sensitive to palmityl-CoA. Our results thus suggest that inhibition by palmityl-CoA may be relatively nonspecific. In this connection, it has been found that the inhibition of the citrate condensing enzyme by palmityl-CoA (Wieland and Weiss, 1963; Tubbs, P. K., 1963) is apparently caused by nonspecific binding of the acyl-CoA to the enzyme (Srere, personal communication).

Rate studies carried out with 1-C¹⁴-acetate, 1-C¹⁴-acetyl-CoA, and 1,3-C¹⁴-malonyl-CoA demonstrated that the preparations employed here incorporate malonyl-CoA into fatty acids approximately 10 times faster than either acetate or acetyl-CoA. This finding suggested that the stimulation of fatty acid synthesis from acetate by citrate or

L-a-glycerophosphate might be at 12.200 and 12.400 and 12.400 and 12.4000.

Accordingly, these compounds were 12.400 and 12.4000 and 12.40

Table IV contains the results of a second of which elevate, fructose-1.6-diphosphate and L-c- β rer proposes the each round to stimulate the carboxylation of acctyl-C and supporting the contention noted above. Figure 1 contains beans for radioactivity in the products formed during this experiment, and shows that in each case the incorporated radioactivity travels as a single peak with the $R_{\rm f}$ of malonyl-CoA.

Summary. Long-chain fatty acid synthesis was demonstrated in high-speed supernatants from yeast. Citrate, glucose-6-phosphate, fructose-1.6-diphosphate, and L-α-glycerophosphate were found to stimulate fatty acid synthesis several-fold. The site of stimulation appears to be the carboxylation of acetyl-CoA. The mechanism of citrate stimulation is not known. Palmityl-CoA inhibited both fatty acid and nonsaponifiable lipid synthesis. However, L-α-glycerophosphate failed to reverse the inhibition, suggesting that glycerophosphate stimulation of fatty acid synthesis is not merely a reversal of acyl-CoA inhibition.

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TABLE I

Effect of Citrate on the Endironation of Red be-Abetical in Long-Continuous Red be-Abetical in State (Section 1988).

Citrate (mM)	Arstate freezlen (mpmoles/nl)		
1.0 2.5 5.0 10.0 25.0 50.0	200 2.4 2.2 2.2 1772		

Incubation mixture contained in a final volume of 2 ml the following: 20 mg supernatnat protein, 150 μ moles potassium phosphate buffer (pH 6.5), 2 μ moles 1-C ¹⁴-acetate (9.6×10⁵ cpm), 5 μ moles ATP, 20 μ moles creatine phosphate, 0.5 mg creatine phosphokinase, 0.5 μ moles NADP⁺, 20 μ moles glucose-6-phosphate, * 0.1 μ mole CoA, 120 μ moles HCO₃, and 5 μ moles MnCl₂. Samples were incubated for 15, 30, and 60 min and the rates of incorporation were estimated from the linear portion of the curves.

*Other experiments have shown that the glucose-6-phosphate keeps the NADP in a reduced form throughout the incubation period.

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TABLL II

Effect of L-α-Glycerophosphate on the Incomposition of 1-C ¹⁴-Acetate Into Long-Chain Fatty Acids and Noncapositionale Lipids

L-\a-Glycerophosphate (mM)	Acetate incorporation (mpsoles)		
(111-1)	Fatty Acids	Nonsaponifiables	
0 0.8 4.0 8.0 16.0	156 532 625 645 711	10 ⁴ 122 83 82 62	

Incubation mixture contained in a final volume of 2.0 ml the following: 20 mg H.S.S., 150 $\mu moles$ potassium phosphate buffer (pH 6.5), 2 $\mu moles$ 1-C 14 -acetate (2×10 cpm), 5 $\mu moles$ ATP, 20 $\mu moles$ creatine phosphate, 0.5 mg creatine phosphokinase, 4 $\mu moles$ NADPH, 0.1 $\mu mole$ CoA, 120 $\mu moles$ HCO3, 5 $\mu moles$ MnCl2; and where indicated, L- α -glycerophosphate. Samples were incubated for 15 min. Acetate incorporation is linear for at least 30 min. under these conditions.

Effect of Palmityl-CoA and Leave cor

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ADDITIONS		EKPERILLET II			
	ACIDS	ÀCH- SAP.	FATTY ACIDD	NON- SAP.	
NONE	<u>5</u>].	15	• 37	36	
L-α-GLYCEROPHOSPHATE lo mM 25 mM	90 -	11	7 ¹ 4 103	27 18	
PALMITYL-COA 0.2 mM 0.6 mM 2.0 mM	3 1 <1	7 < 1 < 1	3 1 -	15 2 -	
0.2 mM PALMITYL-CoA +10 mM L-α-GLYCEROPHOSPHATE +25 mM L-α-GLYCEROPHOSPHATE	2 -	<u>-</u>	1 2	7 2	

Experiment I

Incubation mixtures contained in a final volume of 0.5 ml the following: 6 mg dialyzed H.S.S. protein, 13 μ moles potassium phosphate buffer (pH 6.5), 1.3 μ moles ATP, 5.5 μ moles creatine phosphate, 2.2 units creatine phosphokinase, 0.026 μ moles CoA, 1.5 μ moles MnCl₂, 1 μ moles NADPH, 30 μ moles HCO₃, and 0.5 μ mole 1-C¹⁴-acetate (2×10⁶ cpm/ μ mole). Samples were incubated for 1 hr.

Experiment II

Same incubation mixture as Experiment I except that 1 μ mole of 1-C¹⁴-acetate (2×10⁶ cpm/ μ mole) was used.

Effect of Citrate, Fructose-1.6-. osp. osp. os hea-Chycerophosphate on Acetyl-Co/Carbon osc.

Additions		Bicarbonate Incorporation (cpm)			
None Citrate (60 mM) Fructose-1.6-PO ₄ L-\alpha-Glycero-PO ₄	(30 mM) (60 mM) (3 mM)	+ Acetyl-CoA 14,000 240,000 88,000 152,000 148,000	- Acetyl-CoA 240 240 2000		

Incubation mixtures contained the following: 0-30% ammonium sulfate fraction (2 mg protein), 19 µmoles potassium phosphate buffer (pH 6.5), 2 µmoles ATP, 1 µmole MnCl₂, 5 µmoles creatine phosphate, 2 e.u. of creatine phosphokinase, 1 µmole acetyl-CoA where indicated, 20 µmoles of C¹⁴-bicarbonate (1×10⁶ cpm/µmole), and additions as indicated in the Table. The samples were incubated in a final volume of 0.4 ml for 90 min. at 25 C at which time excess dowex 50w-x8(H⁺) was added to bring the pH to approximately 2. The mixtures were then centrifuged and an aliquot of the supernatant plated and counted.

At the second

Fig. 1. Nonvolatile radioactive discussions of the products forms to the subject of the products forms to the subject of the IV was chromatographed overnight of the subject of the subject of the product of the presence of citrate (5, resistant of the presence of fructose-diPO₄ (4) resistant of the presence of glycero-PO₄.

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(Numbers above 53 not assigned.)

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